高雄醫學大學 105 學年度學士後醫學系招生考試試題

科目:有機化學

考試時間: 80 分鐘

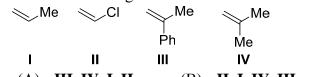
說明:一、選擇題用 2B 鉛筆在「答案卡」上作答,修正時應以橡皮擦擦拭,不得使用修 正液(帶),未遵照正確作答方法而致電腦無法判讀者,考生自行負責。

二、試題及答案卡必須繳回,不得攜出試場。

Choose one best answer for the following questions

【單選題】每題1分,共計60分,答錯1題倒扣0.25分,倒扣至本大題零分為止,未作答,不給分亦不扣分。

Rank the following monomers in order of **increasing** reactivity toward cationic polymerization (least reactive to most reactive).



- (A) III, IV, I, II
- (B) **II**, **I**, **IV**, **III**
- I, II, IV, III (C)
- (D) **IV**, **III**, **I**, **II**
- (E) **I**, **II**, **III**, **IV**

What product would be obtained for the following reaction?

$$CH_3(CH_2)_3C\equiv CMgBr$$
 \xrightarrow{DMF} \uparrow then H_3O^{\oplus}

- (A) CH₃(CH₂)₃C≡CCH₂OH
- (B) CH₃(CH₂)₃C≡CCHO
- (C) $CH_3(CH_2)_3C\equiv CH$

- (D) $CH_3(CH_2)_3C \equiv CNMe_2$
- None of the above.
- Which two have the **same** molecular geometry?
 - I. CO_2
- NO_2^{Θ} II.
- III. PF₃
- SO₄^{2⊖}
- NO₂

- (A) I, II
- (B) III, IV
- I, V (C)
- II, V (D)
- (E) **IV**, **V**

Which nitrogen(s) have **more** basic?

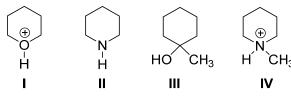
$$I \longrightarrow N \qquad \qquad NMe_2$$

$$\downarrow O \qquad \qquad \parallel \parallel$$

$$H_2N \longleftarrow \parallel$$

- (A) **I**
- (B) **II**
- (C) III
- (D) I, II, and III are acidic
- (E) None of the above.

Rank the acidity of the following compounds.



- (A) $\mathbf{I} > \mathbf{IV} > \mathbf{III} > \mathbf{II}$ (B) $\mathbf{I} > \mathbf{III} > \mathbf{IV} > \mathbf{II}$ (C) $\mathbf{IV} > \mathbf{II} > \mathbf{I} > \mathbf{III}$
- (D) $\mathbf{II} > \mathbf{IV} > \mathbf{III}$
- None of the above.

What is the order of **increasing** acidity for the following compounds?

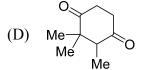


- (A) IV < I < II < III
- (B) III < II < IV
- (C) III < II < IV < I
- (D) IV < II < I < III
- (E) $\mathbf{I} < \mathbf{IV} < \mathbf{II} < \mathbf{III}$

- Which of these substances contains both covalent and ionic bonds?
 - (A) HN_3
- (B) NH_4C1
- (C) H_2O_2
- (D) XeF_2
- (E) PCl₅

Choose the **correct** product of the following reaction?

Me Me
$$\frac{1. O_3}{2. Zn, AcOH}$$
 ?



How many monochloro substituted products C₆H₁₃Cl you might obtain by reaction of 2-methylpentane with Cl₂?

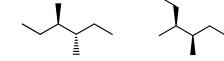
CH₃CH₂CH₂CH(CH₃)₂

- (A) 1
- (B)
- (C) 3
- (D) 4
- (E) 5
- 10. There are some isomers of 4-t-butylcyclhexane-1,3-diol. Which isomer reacts readily with acetone and an acid catalyst to form an acetal, but other stereoisomers do not react?

- 11. Which of the following **correctly** describes a molecule that is achiral?
 - (A) Non-superimposability of the molecule on its mirror image Superimposability of the molecule on its mirror image (B)
 - Contains a carbon atom with four different substituents
- Does not have a plane of symmetry

- Both (B) and (D). (E)
- 12. Which of the following pairs are enantiomers?







- II, IV (A)

(B) I, III

I, II, III

- (D) I, II, V
- (E) **I**, **III**, **V**

13. Consider the two energy diagrams Fig. I and Fig. II given below.

Fig. I



Fig. II



Which of the following is **correct** with respect to these diagrams?

- (A) Fig. I represents an S_N 2 reaction
- Fig. II represents an S_N1 reaction (B)
- (C) Fig. II represents an S_N 2 reaction

- (D) Fig. I represents an S_N1 reaction
- (E) Both (C) and (D).

(C)

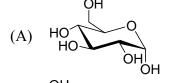
14. Find the energy cost of a 1,3-diaxial interaction for the following compounds, which has most 1,3-diaxial interaction energy? Assume the following 1,3-diaxial strains.

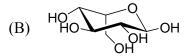
 $CH(CH_3)_2$: 4.2 kJ/mol F: 1.0 kJ/mol CN: 0.8 kJ/mol Cl: 1.0 kJ/mol CH₃: 3.8 kJ/mol

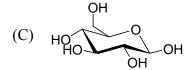
- (A) Isopropylcyclohexane
- (B) Fluorocyclohexane
- (C) Cyclohexanecarbonitrile

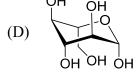
- (D) cis-1-Chloro-2-methylcyclohexane (E)
 - trans-1-Chloro-2-methylcyclohexane.
- 15. Which of the following species is the **least** nucleophilic?
 - (A) Me_3CO^{\bigcirc}
- (B) H_2O
- (C) Me₃N
- (D) BF₃
- (E) [©]CN

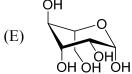
What is the **correct** structure for α -D-glucopyranose?





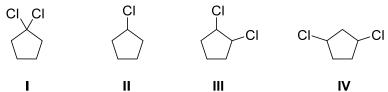






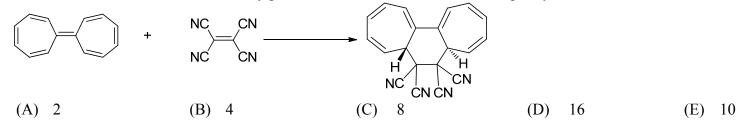
17.	Which of the following	g is a correct repres	entation of the an	nino acid below?				
	H ₂ N-CHC-OH CHCH ₃ CH ₃							
	(A) Isoleucine	resent this amino ac	(B) Ile eid (E) Eithe	r (A), (B), or (C) r	represent	(C) I this amino acid.		
18.	In humans, most steroi (A) Enzymes	ds function as: (B) Hormor	nes (C)	Nucleic acids	(D)	Proteins	(E) Sa	accharides.
19.	What is the IUPAC na Me CHO Me H Me							
	(A) (2 <i>S</i> ,4 <i>R</i>)-Dimeth (D) (<i>S</i>)-2,4-Dimethy	- 1	. , . , ,	Dimethylpentanal Dimethlypentanal.		(C) (R)-2,4-	Dimethyl	pentanal
20.	How many E configura	ation are there in th	e following comp					
	Me CO ₂ H	CHO Br	NH ₂	NC Et		=\		
		Ме	•	Me ₂ N	\sim	Br		
	(A) 1	(B) 2	(C)		(D)	4	(E) 5	
21.	If silver nitrate is conv(A) Exposure to oxy(D) Exposure to alc	ygen	(B) Exposure to		ess for su	ch a conversion i (C) Exposure		:
22.	A compound with the	-	r formula contain	s two double bond	ds. What	is the correct sub	script for	H in the
	formula? C ₁₀ H ₂ ClN ₂ O (A) 19	(B) 22	(C)	18	(D)	20	(E) 21	
23.	When butane undergoe How many times more (A) 100		, 1	action is a seconda		gen in butane tha		mary hydrogen?
24.	(A) Change the read (C) Becomes a perm	g is not a property of ctivity of a function manent part of the p properties of a prot	al group roduct	(B) Iner		ion conditions echanism of the d	lesired rea	action
25.	Which of the following	g carbonyl groups e	xhibits the highe	st wavenumber in	infrared	spectroscopy?		0
	(A)	(B)	(C)	CI	(D)	F	(E) (CI
26.	Compound X has the raphenylhydrazone, bu							
	(A) OH	Иe	(B)	Me		(C) Me	Н	
	(D) Me Me	H	(E)	Et O		IVIC		
27.	For the mass spectrum heights in the ratio of 9 (A) cis-1,2-Dichloro	9 : 6 : 1. What woul	d the compound (B) 2-Chlorope	Y be? entane	n region (the peak velohexane
20	(D) Bromobenzene	11	(E) 1,2-Dibror	•				
28.	Which of the following (A) ² H	g would not produc (B) ¹⁴ N	e nuclear magnet (C)	ic resonance?	(D)	19 F	(E) 11	¹ B

29. Which compounds have 3 signals in the ¹³C NMR spectrum?



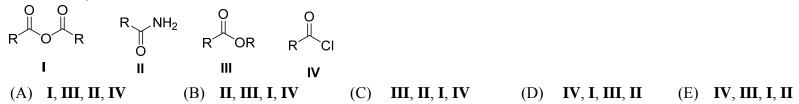
- (A) I, II
- (B) III, IV
- (C) **I**, **II**, **III**
- (D) **I**, **III**, **IV**
- (E) **I**, **II**, **III**, **IV**
- 30. Using a 300 MHz ¹H NMR instrument, if a H shows a triplet at δ 4.02, 4.00, 3.98 ppm, please calculate it's coupling constant. And where will this triplet peak shows up at a 600 MHz ¹H NMR instrument?
 - (A) 6 Hz; δ : 4.01, 4.00, 3.99 ppm
- (B) 6 Hz; δ: 4.02, 4.00, 3.98 ppm
- (C) 6 Hz; δ: δ: 4.00, 3.98, 3.96 ppm

- (D) 2 Hz; δ: 4.02, 4.00, 3.98 ppm
- (E) 2 Hz; δ: 4.00, 3.98, 3.96 ppm.
- 31. Consider the reaction below, how many pairs of electrons are involved in this pericyclic reaction?



- 32. Which of the following reaction types are pericyclic reactions?
 - (A) Diels-Alder reaction
- (B) Cope rearrangement
- (C) Claisen rearrangement

- (D) Stork reaction
- (E) All except (D) are pericyclic reactions.
- 33. What is the order of **decreasing** reactivity towards nucleophilic acyl substitution for the carboxylic acid derivatives? (most reactive first)



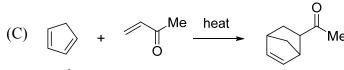
- 34. Consider the elimination reaction: 2-bromohexane was treated with sodium methoxide in methanol. The product(s) of the reaction is(are):
 - (A)

- (B) /
- (C) An equimolar mixture of (A) and (B)
- (D) A mixture of the major product (A) with the minor product (B)
- (E) A mixture of the major product (B) with the minor product (A).
- 35. To answer the following question, consider the reaction below:

The dehydration of alcohol by reaction with POCl₃ in pyridine is an example of:

- (A) E1 process
- (B) $S_N 1$ process
- (C) E2 process
- (D) $S_N 2$ process
- (E) None of above.

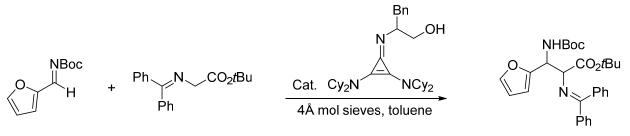
- 36. Which is addition reaction in the following reactions?
 - (A) CH_3CH_2Br + NaCN \longrightarrow CH_3CH_2CN
- (B) OH acid catalyst



(D
$$+ O_2N-NO_2 \xrightarrow{light} NO_2$$

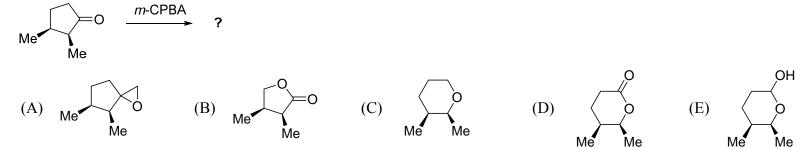
- (E) heat o
- 37. Which of the following substrates will **not** form a Grignard reagent when treated with Mg/diethyl ether?
 - (A) HO
- (B) = Br
- (C) CI
- (D) Br OMe
- (E) Br OMe

38. Which is the name reaction in the following reaction?

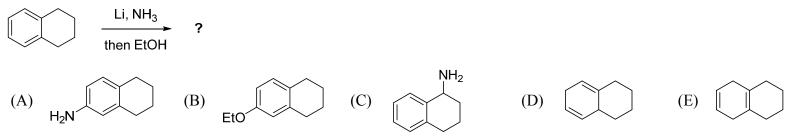


- Mannich reaction
- McMurry reaction
- (D) Dess-Martin reaction
- (E) Dieckmann reaction.
- (C) Wolff-Kishner reaction

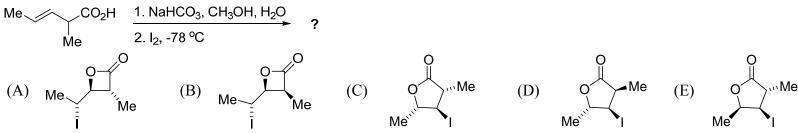
39. What is the **major** product of the following reaction?



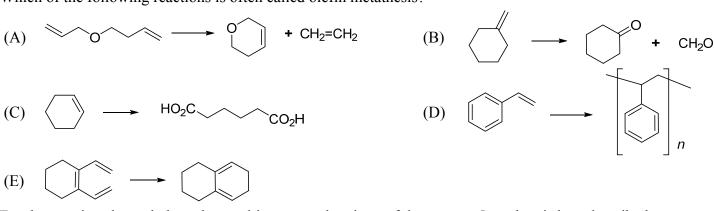
40. Predict the structure of the **expected** product for the following reaction.



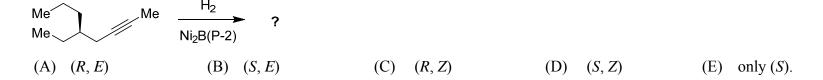
41. What product would be obtained from the following reaction?



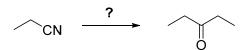
42. Which structure for the compound A (formula C₅H₈O) that fit the following proton NMR data? Chemical shift δ : 1.55 (singlet, 6H), 2.27 (broad singlet, 1H), 2.46 (singlet, 1H) ppm.



44. For the reaction shown below, the resulting stereochemistry of the **expected** product is best described as:

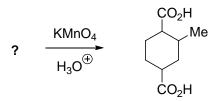


45. How would you prepare the following carbonyl compound from a nitrile?



- (A) 1) EtMgBr; 2) NaOH, H₂O
- (B) 1) EtMgBr; 2) LiAlH₄; 3) H₃O[⊕]
- (C) 1) EtMgBr; 2) H₂O

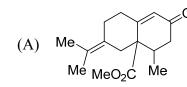
- (D) 2) EtMgBr; 2) CO_2 ; 3) H_3O^{\oplus}
 - I_3O^{\oplus} (E) 1) EtMgBr; 2) PCC
- 46. Which one is the reactant of the following reaction?

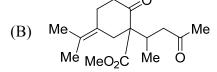


- (A) Me
- (B) M
- (C) Me
- (D) Me
- E) Me

47. Which is the **best** reagent for following reaction?

- (A) PCC
- (B) Jones reagent
- (C) MnO₂
- (D) Ag₂O
- (E) KMnO₄
- 48. An epoxide compound may undergo the ring-opening reaction with water to generate
 - (A) Triol
- (B) Peroxide
- (C) Glycol
- (D) Glycol ether
- (E) Ethylene oxide.
- 49. Show how you might use an annulation reaction to synthesize the following compound. Draw the structure of final product.

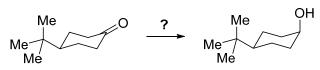




(C)
$$Me$$
 CO_2Me

- 50. Which of the following ethers **can't** be prepared by a Williamson ether synthesis?
 - (A) *t*-Butyl phenyl ether
- (B) Isopropyl methyl ether
- (C) Anisole

- (D) t-Butyl methyl ether
- (E) None of the above.
- 51. Which of the following reagents is **suitable** for the following transformation?



- I. LiAlH₄(A) I
- II. $\text{LiAl}(t\text{-BuO})_3\text{H}$ (B) II
- **III**. LiB(*s*-Bu)₃H (C) **III**
- (D) **I**, **II**
- (E) **II**, **III**

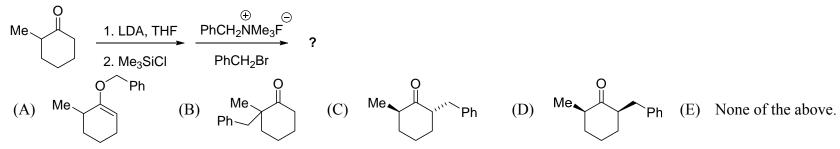
52. What product would be obtained from the following reaction?

- (A) Ph
- (B) Pr
- (C) Ph
- (D) Ph
- E) SMe

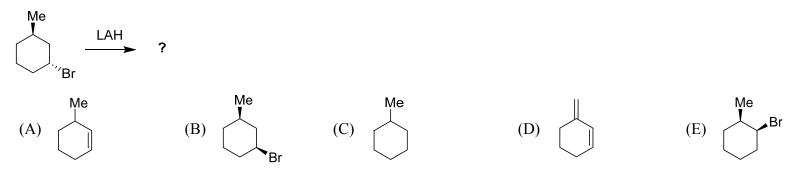
53. What is the product of this reaction?

- (A) Me Me
- (B) Me
- (C) Me CO_2Me
- (D) Me
- (E) CO_2Et

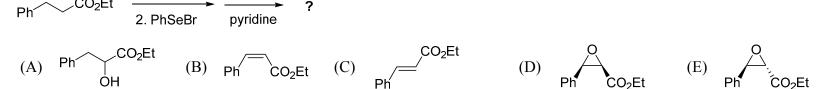
54. What is the **major** product would you obtain for the following reaction?



55. Provide the structure of the **major** organic product in the reaction shown below.

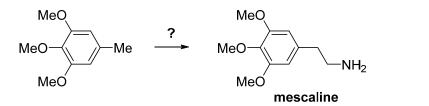


56. What product would be obtained for the following reaction?



57. Please predict the product of the following reaction.

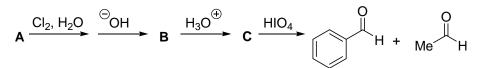
58. Mescaline is a hallucinogenic alkaloid isolated from peyote cactus. Synthesize mescaline from 3,4,5-trimethoxytoluene. Show all reagents toward the target compound.



- (A) 1) NBS; 2) NaCN; 3) LiAlH₄
- (B) 1) *n*-BuLi; 2) BrCH₂NH₂
- (C) 1) Br₂; 2) LiCH₂CN

- (D) 1) light; 2) BrCH₂NH₂
- (E) 1) NBS; 2) LiCH₂NH₂
- 59. Which set of reagents will **best** convert 2,2-dimethylpropan-1-ol to 4,4-dimethylpentan-2-ol?
 - (A) 1) HCl, ZnCl₂; 2) 2 eq. Mg; 3) $CH_2O_{;}H_3O_{;}$
- (B) 1) SOCl₂; 2) 2 eq. Mg; 3) MeCHO; H₃O^{**⊕**}
- (C) 1) SOCl₂; 2) 2eq. Mg; 3) CH₂O; H₃O⁽¹⁾
- (D) 1) HCl, ZnCl₂; 2) 1 eq. Mg; 3) CH_2O ; H_3O^{\oplus}
- (E) 1) HCl; 2) 1 eq. Mg; 3) MeCHO; H_3O^{\oplus}

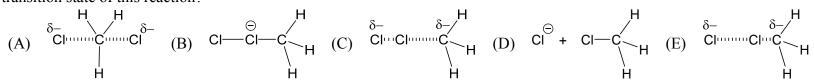
60. Compound **A** can make Br₂/CCl₄ become colorless. What is the structure of compound **B**?



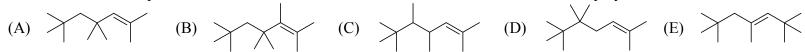
- $(B) \quad \text{Ph} ^{\swarrow} \text{Me}$
- (C) Ph
- (D) Ph Me (E) Ph.

【單選題】每題2分,共計40分,答錯1題倒扣0.5分,倒扣至本大題零分為止,未作答,不給分亦不扣分。

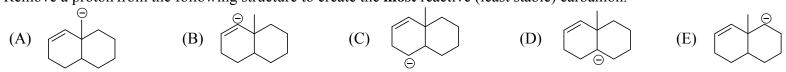
61. The reaction of Cl_2 with a methyl radical has a positive ΔH° . Which of these drawings is the **best** representation of the transition state of this reaction?



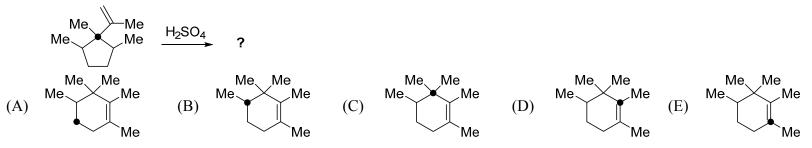
62. Which structure corresponds to the trimer of Me₂C=CH₂ formed under condition of cationic polymerization?



63. Remove a proton from the following structure to create the **most** reactive (least stable) carbanion.



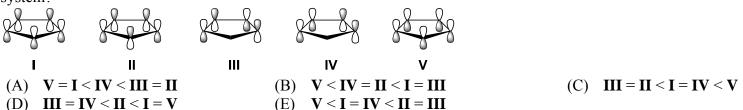
64. The carbon marked by a dot (•) is ¹³C isotope. Which structure below shows the **correct** position of the ¹³C in the product for the carbocation rearrangement shown above?



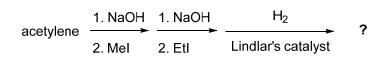
65. Choose substituents X and Y (listed in order below) for the following compound so as to make a Z isomer.

$$X$$
 CN (A) $-Br$, $-NHMe$ (B) $-F$, $-CHO$ (C) $-I$, $-OMe$ (D) $-CO_2H$, $-CH_2NH_2$ (E) $-Br$, $-CO_2H$

- 66. Rank the degree of unsaturation in each of the following compounds.
 - $\textbf{I.} \ Cholesterol\ ,\ C_{27}H_{46}O \quad \textbf{II.} \ DDT,\ C_{14}H_9Cl_5 \quad \textbf{III.} \ Prostaglandin\ E1,\ C_{20}H_{34}O_5 \quad \textbf{IV}.\ Caffeine\ ,\ C_8H_{10}N_4O_2$ $(A) \quad I > III > II > IV$ (B) $\mathbf{II} > \mathbf{IV} > \mathbf{I} > \mathbf{III}$ $(C) \quad \mathbf{I} > \mathbf{II} > \mathbf{III} > \mathbf{IV}$ (D) $\mathbf{II} > \mathbf{IV} > \mathbf{III} > \mathbf{I}$ (E) $\mathbf{I} > \mathbf{IV} > \mathbf{II} > \mathbf{III}$
- 67. This cyclic carbocation has two sets of degenerate *pi*-molecular orbitals. Choose the **correct** order MO's energies for this system?



68. What product would be obtained for the following reaction?



(B) Me Et (D) (E) None of the above.

- 69. When 1-methyl-1-cyclohexene is respectively treated with the following reagent set, which will give the **same** product?
 - I. 1) BH₃, THF; 2) H₂O₂, NaOH, H₂O III. 1) *m*-CPBA; 2) H₃O ⊕

II. 1) Hg(OAc)₂, H₂O, THF; 2) NaBH₄

IV. 1) OsO₄; 2) NaHCO₃, H₂O

- \mathbf{V} . $\mathbf{H}_{3}\mathbf{O}^{\mathbf{\oplus}}$
- (A) II, V
- (B) **III**, **IV**
- II, III, V
- (D) I, V
- (E) **I**, **III**

70. Predict the outcome of the following sequence of reactions.

$$Me = CO_2Me \xrightarrow{1. HSCH_2CO_2Me}$$
2. NaOMe

71. The following substrate is a starting material in the synthesis of compounds having opioid activity. Show all products that would result from the reaction below.

72. Predict the structure of the hydrolysis product.

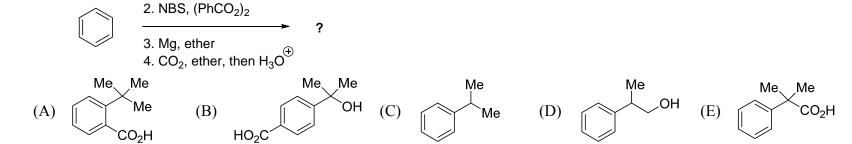
73. In order to synthesize the final product shown below, predict when should a protecting group be added and when should it be removed?

- (A) The hydroxyl group should be protected before step \mathbf{I} and removed after step \mathbf{I}
- (B) The hydroxyl group should be protected before step ${\bf I}$ and removed after step ${\bf II}$
- (C) The hydroxyl group should be protected before step I and removed after step III
- (D) The hydroxyl group should be protected before step II and removed after step III
- (E) There is no need for a protecting group in this synthesis.
- 74. Which of the following Diels-Alder reactions has the **largest** reaction rate constant?

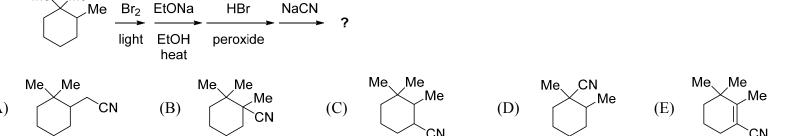
$$(A) \quad \stackrel{\mathsf{CN}}{\longleftarrow} \quad \mathsf{CHO} \quad (B) \quad \stackrel{\mathsf{Me}}{\longleftarrow} \quad \mathsf{(C)} \quad \stackrel{\mathsf{CHO}}{\longleftarrow} \quad \mathsf{(D)} \quad \stackrel{\mathsf{OMe}}{\longleftarrow} \quad \mathsf{CHO} \quad (E) \quad \stackrel{\mathsf{OHC}}{\longleftarrow} \quad \mathsf{Me} \quad \mathsf{CHO} \quad \mathsf{(E)} \quad \mathsf{(D)} \quad \mathsf$$

75. What is the product of this reaction?

1. Me₂CHCl, AlCl₃



76. What is the **major** product obtained from the following reaction sequence?



77. What is the **major** product of the following reaction?

$$(A) \xrightarrow{\mathsf{Br}} (B) \xrightarrow{\mathsf{Br}} (C) \xrightarrow{\mathsf{Br}} (D) \xrightarrow{\mathsf{Br}} (E) \xrightarrow{\mathsf{Br}}$$

78. Which of the following reactions is called Gabriel synthesis?

(E) None of the above.

79. What is the **major** product of the following triene to undergo the intramolecular Diels-Alder reaction?

80. The following reaction involves an intramolecular Michael reaction followed by an intramolecular aldol reaction. What is the **major** product of this reaction?

後醫-英文

題號	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20
答案	C	Е	A	C	Е	A	В	D	Е	A	C	В	A	D	В	A	D	A	A	A
題號	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40
答案	A	C	C	D	В	A	В	A	E	A	D	A	В	В	E	В	C	A	D	В
題號	41	42	43	44	45	46	47	48	49	50										
答案	В	C	В	D	A	C	В	A	D	C	·		·							

後醫-有機化學

DATI TANKING A																				
題號	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20
答案	В	В	C	C	A	В	В	В	Е	A	В	C	E	E	D	A	E	В	D	C
題號	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40
答案	D	A	В	C	D	В	A	C	Е	A	C	Е	D	D	C	C	A	A	D	Е
題號	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60
答案	C	В	A	D	C	A	C	C	A	A	C	В	A	C	C	C	В	A	В	A
題號	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80
答案	Е	A	D	D	В	В	Е	Е	A	В	В	D	В	D	Е	C	D	C	C	A

後醫-普通生物學

題號	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20
答案	C	D	A	Е	A	C	D	Е	В	A	В	В	D	A	A	C	Е	Е	В	Е
題號	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40
答案	D	В	A	В	C	A	D	D	A	D	E	C	D	C	C	E	A	В	C	C
題號	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60
答案	В	A	В	D	В	В	C	A	C	Е	A	Е	D	D	A	A	В	C	Е	D
題號	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80
答案	D	E	C	C	В	E	C	A	В	D	В	Е	В	E	В	D	E	A	В	E